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Ultrafast Optical Study of Small Gold Monolayer Protected Clusters: A Closer Look at Emission†

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Monolayer-protected metal nanoclusters (MPCs) were investigated to probe their fundamental excitation and emission properties. In particular, gold MPCs were probed by steady-state and time-resolved spectroscopic measurements; the results were used to examine the mechanism of emission in relation to the excited states in these systems. In steady-state measurements, the photoluminescence of gold clusters in the range of 25 to 140 atoms was considerably stronger relative to larger particle analogues. The increase in emission efficiency (for Au25, Au55, and Au140 on the order of 10−5) over bulk gold may arise from a different mechanism of photoluminescence, as suggested by measurements on larger gold spheres and rods. Results of fluorescence upconversion found considerably longer lifetimes for smaller gold particles than for larger particles. Measurements of the femtosecond transient absorption of the smaller clusters suggested dramatically different behavior than what was observed for larger particles. These results, combined with the result of a new bleach band in the transient absorption signal (which is presumably due to an unforeseen ground state absorption), suggest that quantum size effects and associated discrete molecular-like state structure play a key role in enhanced visible fluorescence of small clusters.

I. Introduction

Metal nanoparticles have captured the attention of the research community in the past decade with their tunable physical and chemical properties.1–8 These materials hold great potential in a wide range of applications, including highly efficient catalysts,9–15 medical imaging markers,16,17 and various molecular electronics.4,9,18,19 There has been a great deal of optical and electronic investigation of the specific properties of nanoparticles at various sizes. Bulk metal, metal thin films, nanoparticles, as well as monolayer protected metal clusters (MPCs) have been reported in the literature. Typically, bulk metal has a system size that is larger than the wavelength of the external electromagnetic field, and its properties can be described with the dielectric functions of the metal. For metal nanoparticles, where the core size is between 3 and 50 nm in diameter, there is a strong dipolar SPR. The resonance for gold nanoparticles is in the visible region. The majority of the previous optical and detailed electronic characterization studies of metal nanostructures had been focused on the nanoparticles and nanoclusters.1–8,14–20 Metal clusters had been investigated previously in the gas phase;2,20 with the discovery of metal clusters in the condensed phase, a great deal of interest is developed in the application, properties, and fundamental physics of these nanosystems.1–8,14–25

The interest in nanoparticles by a large following of scientists and engineers focuses on the properties and applications of small metal topologies. In particular, there is a great interest in the possibility that these small clusters of metal (<2.5 nm) can be used in catalysis,10–15 linear and nonlinear optical effects,26–28 and “quantized” electrochemical applications.4,9,18,19 Although some applications of metal clusters have been well researched, the fundamental science regarding the structure and electronic states are still under debate. There have been a number of different approaches to make small metal clusters. For example, Tomalia et al.29,30 reported the first synthesis of smaller gold particles under the context of making dendrimer metal nanocomposites. It was later found with these systems that tremendous optical limiting and time-resolved optical properties31–33 could be obtained as well as new ideas for catalysis.9–15 Highly stable systems, such as gold, became a major focus for new nanosystem synthesis.34,35 The search for even smaller gold particles made in this regard was reported by Crooks et al.36,37 Chandler et al.38,39 and Dickson et al.7 utilizing the dendrimer capture approach with PAMAM. Also, Bauer et al.40 utilized a G9 dendrimer (PAMAM) and reported a nanoparticle with size of ~1 nm as well. Optical studies conducted on this particular dendrimer small metal particle system suggested that the emission (fluorescence) mechanism might vary depending on the size of the metal particle (cluster).33 Other synthetic routes were also explored using a ligand-protected approach first explored by the Schmid group.5,41 Subsequent progress in the synthetic methods,42–46 especially by the Brust group,45 has led to well-isolated Au nanoparticles of various distinct sizes47 and the production of soluble isolated clusters in large quantities. It was suggested that the nanoparticles produced by the Brust synthesis (and its variations) contain a single layer of ligand shell, leading to the name monolayer protected clusters (MPCs).1,3,5,8 Gold nanosystems can be divided into two classes, nanoparticles (>2 nm) and nanoclusters (<2 nm), separated by...
size and the presence of surface plasmon resonance (SPR) in the absorption spectrum. The progress in synthesis and purification paved way to the breakthrough of a total structural determination of Au nanoclusters $\text{Au}_{10} \left( \text{SR} \right)_{4}$ and $\text{Au}_{25} \left( \text{SR} \right)_{18}$ through X-ray crystallography. A number of analytical techniques were also used to understand the composition of the clusters, including HR-TEM, small-angle X-ray scattering, XPS, and thermogravimetric methods. To understand the physical sizes and distribution of nanosystems, TEM was most commonly used. Identifying clusters using size alone, $\text{Au}_{30}$, $\text{Au}_{55}$, $\text{Au}_{100}$, and others were labeled. As the synthesis techniques advanced, the increase in stability and purity of the samples has led to more sophisticated analytical techniques. In particular, mass spectroscopy proved to be an indispensable tool, identifying not only the core mass but also the ligand shell ratio. $\text{Au}_{25} \left( \text{SR} \right)_{18}$, $\text{Au}_{30} \left( \text{SR} \right)_{24}$, and $\text{Au}_{55} \left( \text{SR} \right)_{39}$ were identified using mass spectroscopy. For this article, the clusters will be labeled only with by the core size.

Optical and electrochemical properties have recently been the focus of small metal nanoparticle studies. The emission (fluorescence) properties of a number of gold particle topologies has been investigated. In many studies; the complexity of nanoclusters and the lack of a perfect structural determination, the increase in stability and purity of the samples has led to more sophisticated analytical techniques. In particular, mass spectroscopy proved to be an indispensable tool, identifying not only the core mass but also the ligand shell ratio. $\text{Au}_{25} \left( \text{SR} \right)_{18}$, $\text{Au}_{30} \left( \text{SR} \right)_{24}$, and $\text{Au}_{55} \left( \text{SR} \right)_{39}$ were identified using mass spectroscopy. For this article, the clusters will be labeled only with by the core size.

### Experimental Section

#### A. Sample Preparation

$\text{Au}_{155}$ was prepared by the previously reported methods. Dodecane-functionalized generation five PAMAM dendrimers were used to capture gold from a AuCl$_4^-$ solution. The complexed Au cations were subsequently reduced with NaBH$_4$ in pH 9.0 base. Control of the PAMAM generation and metal/dendrimer ratio allows for size control of the templated particles. The dendrimer-encapsulated Au nanoparticles were extracted in an aqueous solution containing tiopronin. The extraction results in $\text{Au}_{155}$ nanoclusters with tiopronin as the outer shell. The sample was initially characterized by UV−visible spectroscopy spectrum, specifically looking for the lack the SPR and the presence of small features in the spectrum.

The synthesis and characterization of other samples used in our experiments such as $\text{Au}_{25}$, $\text{Au}_{40}$, $\text{Au}_{30}$, $\text{Au}_{76}$ and $\text{Au}_{406}$ were published elsewhere. In brief, metal salts in solution, in this case gold, are reduced with a solution of organic thiols and a phase transfer reagent under temperature-controlled condition to produce MPCs. Typically, a number of analytical techniques such as HR-TEM, small-angle X-ray scattering, XPS, and thermogravimetric analysis are used to characterize MPCs. The sample used in this article was prepared with the same synthesis procedure. Particular samples used in this article were characterized using TEM, where size of the clusters is used to estimate the number of gold atoms and ligands. The assigned core numbers and ligands were derived from the narrow size distributions. Using the results of other characterization recently available, the assigned core number is not perfectly accurate for relatively large clusters with core diameter $\geq 2$ nm. For small clusters, recent studies provided accurate compositions.

#### B. Steady-State Absorption and Emission

Steady-state absorption and emission measurements were performed at room temperature, with a 1 mm thick quartz sample cell. All samples were kept in their powder form, under refrigeration, and were dissolved in the appropriate solvent just before the experiments were performed. Optical absorption measurements were carried out using an Agilent 8432 UV−vis absorption spectrometer. Fluorescence measurements were carried out using a Shimadzu RF-1501 spectrofluorophotometer. The quantum yields of the molecules were measured using Coumarin 307 as the standard. To ensure aggregation did not affect the samples, UV−vis spectra were compared before and after the experiment to ensure the purity of the sample. The calculated absorption spectrum utilized the calculation method based on our previous paper following the Gans extension of the Mie theory. The dielectric constant of the surrounding medium was taken to be 2, and the simulated metal system is considered to be spherical with a diameter of particle is 1.1 nm ($\text{Au}_{25}$). The spectrum is calculated from 390 to 800 nm at 10 nm intervals.

#### C. Femtosecond Transient Absorption Measurements

Transient absorption was used to investigate the excited-state dynamics of metal nanoclusters; the experimental setup has been...
previously described. In brief, the output of an amplified laser beam was split with a beam splitter to generate pump and probe beam pulses (85 and 15%). The pump beam was produced by optical parametric amplifier (OPA-800c). The pump beam used in the present investigation (390 nm) was obtained from the fourth harmonic of the signal beam and was focused onto the sample cuvette. The probe beam was delayed with a computer-controlled motion controller and then focused into a 2 mm sapphire plate to generate white light continuum. The white light was then overlapped with the pump beam in a 2 mm quartz cuvette containing the sample, and the change in absorbance for the signal was collected by a CCD detector.

### III. Results and Discussion

#### A. Steady-State Absorption

The steady-state absorption spectra are shown in Figure 1 for Au25, Au55, Au140, and Au2406. Because of differences in the size and structure of the nanoclusters and nanoparticles, they have different steady-state absorption spectra, namely, the presence of SPR. The surface plasmon is caused by the collective response in the polarization and restoration of conductive electrons and can be observed as a peak in the visible spectrum centered at ∼520 nm for gold nanoparticles. However, SPR is observed only in nanoparticles and not in nanoclusters, and can be used to differentiate between the two systems. From Figure 1, it can be clearly seen that SPR is observed for Au2406, and the lack of SPR for Au25, Au55, and Au140 confirms that the samples are nanoclusters.

Between the nanoclusters, the difference in core size creates a difference in the electronic structure. As it was reported by Jin et al., the three distinct features in the absorption spectrum for Au25 can be correlated to the proposed icosahedra structure. In another report, the absorption spectrum is directly affected by the geometric and electronic structure of the system. Utilizing a calculation method based on Mie theory (as was done previously for larger particles), the absorption spectrum for a gold system similar in size to Au25 can be simulated. The Mie theory predicts the appearance of a surface plasmon band that is not observed in the experimental results; this suggests that Mie theory does not apply for nanoparticles at very small sizes. Comparing the absorption spectrum of Au140 to that of Au55 or Au25 also offers evidence of a quantum size effect. The strength of the characteristic absorption features increases as the size of the cluster decreases, which suggests that the molecular characteristics may possibly be associated with the size of the core.

#### B. Fluorescence

We measured fluorescence from samples excited at 390 nm and found visible emission spectra from the tiopronin-capped Au55 samples with a peak centered at 500 nm. The emission was excited in the range of 360–390 nm and the emission peak from each excitation wavelength was detected at 500 nm. With the aid of time-resolved fluorescence upconversion, the emission decay was found to be ∼250 fs, longer than the IRF of the setup (Figure 3). The emission of Au55 has been an area of interest since its discovery. Crooks et al. studied dendrimer and other synthetic materials used in the synthesis of Au55 nanoclusters but found them to be nonfluorescent. Au55 emission was also investigated by the Murray group in an Ag exchange reaction. Their experiment started with Ag MPCs, monitoring the emission as Ag is exchanged for Au. We observed 500 nm emission for Ag MPCs and in the steps of the reaction but not for Au. The Au55 sample that was used in our experiments was produced through PAMAM dendrimer exchange reaction process. We observed both steady-state and time-resolved emission from Au55. Fluorescence from Au25 and Au140 has been observed.
The quantum yield (QY) for visible emission of Au55 is significantly larger than the 10^{-10} reported for bulk gold.\textsuperscript{27} Compared with the estimated QY for Au25 and the experimental result from Au140, the QY for Au55 is comparable in the visible (Table 1). The infrared QY for Au55 is still under investigation; from the observed trend found for Au25, the QY in the NIR for Au55 could be even stronger than the visible emission.\textsuperscript{27,58,73}

C. On the Mechanism of Emission. From the emission data, we can identify two luminescence transitions: transition from a short-lived state in the visible range, which is most probably associated with the MPC's core,\textsuperscript{21,22,59,64} and that from a long-lived surface-related state in near-infrared most probably associated with ligands.\textsuperscript{21,22,59,64} Our time-resolved fluorescence data showed that the visible emission lifetime (∼250 fs, Figure 3) is substantially longer than the lifetime of the visible emission of larger nanoparticles.\textsuperscript{59} The short-lived visible emission mechanism can be explained by the creation of the hole by the excitation in the ground state, which can be subsequently filled with the electron from the higher lying excited state near 2.5 eV (Figure 4, B band). This transition is similar to the HOMO−LUMO+1 suggested on the basis of crystal structure for Au35(SR)\textsubscript{18}.\textsuperscript{21,22} HOMO−LUMO transition corresponds to the transition from ground state to A band on the diagram shown in Figure 4. The energy gap of this transition is lower than that for the observed visible fluorescence. On the basis of the absence of the detected dynamic Stokes shift, it was suggested that the energy is quickly transferred for A band to the long-lived surface-related (semiring) states from which the NIR fluorescence originates.\textsuperscript{21,22} For larger nanoparticles, the photogenerated d-hole is filled by the s-p electron near the Fermi level, which leads to very short-lived (hole lifetime ∼20−30 fs\textsuperscript{61}) visible emission.\textsuperscript{62} The proximity of surface plasmon in terms of energy can enhance the fluorescence, thus leading the higher quantum yield as compared with bulk metal.\textsuperscript{59}

Surface states were used in reported literature\textsuperscript{22,57,76} to explain the longer fluorescence lifetime in the NIR for nanoclusters. It was suggested that the A band to ground-state transition is mostly nonradiative and does not produce detectable fluorescence. The observed NIR emission is most probably the transition from the surface state to the ground state. Furthermore, the surface state should be affected by the surface type (ligand), and the surface state should be independent of the core size.

We obtained fluorescence up-conversion lifetimes by fitting the experimental profile with multieponential decay functions convoluted with the IRF. Comparison of the fluorescence life times between nanoclusters (Au55, Au25,\textsuperscript{27} Au140,\textsuperscript{27} Au309\textsuperscript{27}) and nanoparticles\textsuperscript{59} (Figure 5) shows that the lifetime for nanoclusters is much longer than that for nanoparticles. The difference in the lifetime could be attributed to the fast filling of the d-holes for the nanoparticles, whereas for nanoclusters, a molecular-like discrete energy level structure results in a slower relaxation process. The distinct difference in the lifetime of nanoclusters and nanoparticles is another manifestation of the quantum size effect, showing that as the particle size changes the emission mechanism also changes. Upon a closer inspection of Au55 and Au976 (Figure 3), it can be seen clearly that the lifetime for Au976 is much shorter than that for Au55. Looking at the relationship between quantum yield and fluorescence lifetime, if we assume that both Au25 and Au976 are close in oscillator strength, we can suggest that there will be a correlation between quantum yield and fluorescence lifetime. This relationship is now observed experimentally.

D. Wavelength-Resolved Transient Absorption. Transient absorption allows for studying excited-state dynamics of the chemical system and gives information about the energy states of the system. The transient absorption of nanoparticles and nanoclusters has been previously studied by various groups.\textsuperscript{26,27,31,64,74–81} Our group has also studied the degenerate transient absorption of Au25 and Au140.\textsuperscript{27,64} In this report, we extend our transient absorption study to wavelength-resolved transient absorption; we have also included Au55 in our investigation. Transient absorption spectra of Au25, Au55, and Au140 were compared at time delay of 550 fs (Figure 6). All three spectra showed no bleaching at 530 nm, an additional indication of the absence of SPR, which correlates with the steady-state absorption data. For Au55 and Au76, the peak at 500 nm shows transitions similar in magnitude, whereas Au140 shows a stronger transient absorption at this wavelength. For the transient absorption at 650 nm, the comparison of Au140, Au55, and Au25 with the same concentration shows that as the core size decreases, the transient absorption signal also decreases. The kinetic data at 640 nm for Au55 (Figure 7) show an initial quick relaxation to the intermediate state and then a slow decay back to the ground state. The dynamics are very different from those associated with SPR transients following electron−phonon relaxation in large particles.\textsuperscript{77} The decay profile is typical for molecular-like small clusters with single electron process.

E. Pump Power Dependence of Transients. Varying the pump power of transient absorption measurements can give information about electron−electron scattering and electron−phonon relaxation for some nanosystems.\textsuperscript{77−81} It has been reported for gold nanoparticles that the electron−phonon relaxation depends strongly on pump power,\textsuperscript{59,98,91} whereas the signal of electron−electron scattering is only weakly power-dependent.\textsuperscript{25} Electron−electron scattering for the gold nanoparticles can be attributed to the sharing of energy by excited electrons and the thermal relaxation of the electronic gas.\textsuperscript{76−79} For electron−phonon relaxation, it corresponds to the transfer of energy from electron system to the lattice.\textsuperscript{77−81} For nanoclusters, the decrease in the size and the presence of quantum size effects (evident in the absorption spectra) suggests that the simple interpretation of the decay in terms of e/e and e/ph processes similar to bulk metal do not apply. We fit the decay with experimental function for the 2.5 nm particle and Au55; we can see that there is a lack of power dependence for Au55 when compared with that of the nanoparticle (Figure 8). The difference in power dependence suggests that the relaxation of the system is affected by the discrete energy levels of a molecule-like system but not the thermal relaxation of the electron−phonon system similar to bulk metal.
F. Bleach Structure. The energy structure for nanocluster Au_{25} has been studied by various groups. Zhu et al.\textsuperscript{21} performed time-dependent density functional theory calculations based on structural information and correlated the calculated result to the experimentally measured absorption spectrum. Moran et al.\textsuperscript{22} used transient absorption experiments to investigate the relaxation processes in Au_{25}(SR)\textsubscript{18}. The analysis of the transient absorption data in this work\textsuperscript{22} showed that the bleaching observed at \(\sim 680\) nm corresponds to a HOMO–LUMO absorption peak and can be assigned to the ground-state bleach. Transient absorption spectrum accessible by our setup covers the range from 450 to 750 nm, which is broader than the 620–800 nm range accessible by Moran et al.\textsuperscript{22} In our experiments, we also observed the dip in transient absorption spectrum near 680 nm (Figure 9). Because of extended spectral range, we were also able to detect another dip near 550 nm (Figure 9). Using similar analysis,\textsuperscript{22} we can suggest that the weakly negative dip in transient absorption spectrum near 550 nm corresponds to the bleach of another transition from the ground state near 2.25 eV due to the ground-state population depletion. This transition could be correlated to the weak shoulder observed in the steady-state absorption spectrum (Figure 10).

IV. Summary

In summary, the emission for Au_{55} has been observed in the visible range. It was time-resolved on a subpicosecond time scale using fluorescence upconversion. The visible emission from Au_{55} is much stronger than that observed for larger nanoparticles. The mechanism of emission was investigated and was found to be different from that for larger nanoparticles and bulk gold. In particular, the existence of molecular-like emission with considerably stronger intensity became evident for the Au_{55} system. This provides stronger evidence of the presence of quantum size effects, which were observed by comparing absorption spectra and fluorescence decay profiles for nanoparticles and nanoclusters. The transient absorption dynamics of Au_{55} was found to be nearly independent of the pump power, which was very different from those observed for larger nanoparticles. Close inspection of transient absorption spectra, and the steady-state absorption of Au_{25} suggests the existence of another state.
not previously considered. The valley at ~550 nm from the transient absorption spectrum may correspond to the bleach of the transition to this state from the ground state due to the ground-state depletion, similar to that observed at ~680 nm.

References and Notes

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