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Designed guanidinium-rich amphipathic oligocarbonate molecular transporters complex, deliver and release siRNA in cells

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The polyanionic nature of oligonucleotides and their enzymatic degradation present challenges for the use of siRNA in research and therapy; among the most notable of these is clinically relevant delivery into cells. To address this problem, we designed and synthesized the first members of a new class of guanidinium-rich amphipathic oligocarbonates that noncovalently complex, deliver, and release siRNA in cells, resulting in robust knockdown of target protein synthesis *in vitro* as determined using a dual-reporter system. The organocatalytic oligomerization used to synthesize these co-oligomers is step-economical and broadly tunable, affording an exceptionally quick strategy to explore chemical space for optimal siRNA delivery in varied applications. The speed and versatility of this approach and the biodegradability of the designed agents make this an attractive strategy for biological tool development, imaging, diagnostics, and therapeutic applications.

amphipathic co-oligomers | nanoparticles | oligonucleotide delivery | biodegradable oligomers | organocatalysis

RNA interference (RNAi) is an emerging technology that is revolutionizing many strategic approaches to biochemical pathway analysis, drug discovery, and therapy (1–6). As part of the RNAi pathway, small interfering RNAs (siRNAs) induce post-transcriptional, sequence-specific gene silencing utilizing endogenous intracellular machinery to selectively suppress gene expression and, thereby, reduce target protein synthesis (7). The net effect is equivalent to protein inhibition without the use of small molecule inhibitors. The specificity of RNAi also allows one to make inhibitors against previously undruggable targets. Both the ubiquity of the RNAi pathway within the body and the ease with which siRNA can be used to suppress a specific target of interest have made siRNAs a promising class of molecules for the treatment of cancer, viral infections, ocular disorders, and genetic diseases (5). In 2004, the first siRNA-based therapy entered Phase 1 clinical trials (4). Since then, several other RNAi-based therapies have reached clinical evaluation for a number of indications including cancer, viral infections, and genetic skin disorders (5, 8, 9). Notwithstanding this progress, formidable challenges remain for the application of RNAi technology in basic research and therapy, the most fundamental of which is delivery of siRNA across biological barriers.

The siRNAs are double-stranded RNA molecules typically consisting of a 19–23 base-paired region with two 3′ overhanging nucleotides. It is polyanionic, polar, and large (*ca.* 13 kDa), compared to small molecule therapeutics. These physical properties suppress or prevent its unassisted passage through nonpolar membranes and, thus, its access to the intracellular RNA-induced silencing complex (RISC) components required for target protein knockdown (6). This problem is further exacerbated by siRNA's susceptibility to enzymatic degradation (*i.e.*, RNases) (3). To address these problems, two strategies have been pursued: development of noncharged and nonbiodegradable siRNA surrogates

(10) and, more directly, development of delivery vehicles and strategies that would enable or enhance the entry of siRNA itself. Several siRNA delivery technologies have been reported thus far, including direct covalent conjugation of siRNA to lipids, peptides, or to aptamers; and noncovalent complexation of siRNA with polymers, biopolymers, nanotubes, lipid-based vehicles (*e.g.*, lipopolyplexes, stable nucleic acid lipid nanoparticles), cyclodextrin polymer-based nanoparticles, fusion proteins, membrane translocation-modified magnetic nanoparticles, and antibody—protamine conjugates (4, 6, 11–22).

In 2000, we reported an extensive reverse engineering effort directed at the highly cationic HIV-Tat 9-mer peptide (RKKRRQRRR), showing that its ability to enter cells is related to its arginine content and, more specifically, to the number and array of its guanidinium groups (23). This finding led to the design of oligoarginine and guanidinium-rich peptoid cell penetrating agents and, subsequently, a wide range of designed non-peptidic agents, more generally and accurately dubbed molecular transporters, differing in backbone structure but uniformly incorporating the key guanidinium head groups (24). These more synthetically accessible homo-oligomeric transporters performed as well as and often better than the hetero-oligomeric Tat 9-mer *in vitro* and *in vivo* studies. We and others have since shown that these guanidinium-rich molecular transporters can enable or enhance the delivery of a variety of cargos, including small molecules, metals, imaging agents, peptides, plasmids, and proteins across biological barriers, such as cell membranes and the stratum corneum, the latter as part of a clinical trial (24, 25). Recently, we developed an oligomerization strategy that generates unique guanidinium-rich homo-oligocarbonate molecular transporters in an exceptionally step-economical fashion (one to two steps) through a metal-free organocatalytic ring-opening oligomerization reaction (26). With this strategy, the length of the oligomer (degree of polymerization, DP) can be easily tuned by varying the ratio of monomer to initiator in the oligomerization step. The guanidinium-rich oligocarbonate molecular transporters have narrow molecular weight distributions ($M_w/M_n = 1.1 - 1.2$), are reproducibly formed over a range of scales (50 mg to 2.5 g), and, like oligoarginines, readily enter cells. Significantly, this oligomerization strategy can also be deployed to produce co-oligomers of widely varied composition and therefore properties, again in

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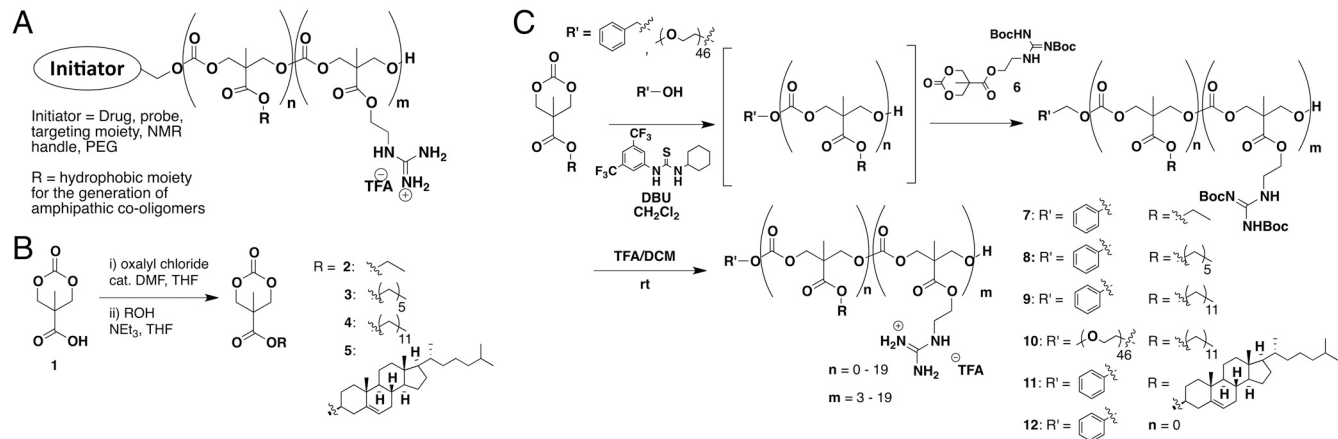


Fig. 1. General structure and synthesis of the guanidinium-rich amphipathic carbonate co-oligomers. (A) General structure of the amphipathic carbonate block co-oligomers. (B) Synthesis of the hydrophobic monomers from carboxylic acid **1**. (C) Amphipathic block co-oligomers were synthesized by organo-catalytic ring opening oligomerization of the monomers, followed by guanidine deprotection with TFA. The scheme depicts the synthesis for a block co-oligomer. Statistical co-oligomers were synthesized by adding both monomers simultaneously at the start of the reaction.

one to two steps through the use of two or more monomers incorporating different side chains (27–29).

We reasoned that the polycationic nature of guanidinium-rich oligocarboxonates, if mixed with additional functionality for cell penetration, could be exploited to noncovalently complex and deliver oligonucleotides into cells. More specifically, we hypothesized that amphipathic carbonate co-oligomers, composed of guanidinium-rich side chains to bind siRNA through electrostatic and hydrogen-bonding associations and hydrophobic side chains to facilitate packing and cellular entry (Fig. 1A), would serve as effective siRNA complexation and delivery vehicles. Requiring only one to two steps to prepare, the metal-free syntheses of these co-oligomers are uniquely short and facile, thereby allowing one to rapidly synthesize, test, and tune co-oligomers for uptake and offering a distinct advantage over lengthy stepwise synthesis. This co-oligomerization strategy would also be readily amenable to the introduction of targeting elements as the initiator moiety. A further advantage of this approach is that these new carbonate co-oligomers would be biodegradable, a feature that could serve to affect both cargo release and transporter clearance. The method of synthesis, the length of these co-oligomers, the use of the guanidinium group as a cationic complexing moiety and, notably, the biodegradable nature of the oligocarboxonate backbone are rare in the siRNA delivery field, and their combination is unique. The distinct advantages offered by this strategy and the speed with which information on siRNA complexation and delivery could be acquired prompted the investigation described herein. We report the study of the synthesis and evaluation of a series of carbonate co-oligomers designed to systematically probe the functionality and factors required for effective complexation, delivery, and release of siRNA.

Results and Discussion

Synthesis of the Guanidinium-Rich Amphipathic Carbonate Co-oligomer Transporters. The initial target set of carbonate co-oligomers was selected to explore the influence of structural variables (e.g., molecular weight, hydrophobicity, ratio of lipid to guanidinium content) on complexation and cellular uptake. As these are the first guanidinium-rich amphipathic co-oligocarbonate transporters to be studied, information on the optimal lipid side chain for siRNA delivery was not known but was expected to be rapidly addressable through the systematic use of monomers incorporating simple, stable and biocompatible lipid side chains (ethyl, hexyl, or dodecyl, Fig. 1B: 2–4). Cholesterol-incorporating monomer **5** was also included to explore the role of more complex polycyclic lipids in both co-oligomer synthesis and performance. Further exemplifying the flexibility and speed of this strategy, the mono-

mers were efficiently accessed by conversion of the cyclic carbonate carboxylic acid **1** to an acid chloride, followed by esterification with selected lipid alcohols (30).

Co-oligomerizations were then conducted using both the requisite hydrophobic monomer and the previously reported guanidine-protected monomer **6** (Fig. 1C) (26). A range of DPs (oligomer lengths) was targeted including “short” (Table 1, $m = n$ = approximately 4), “medium” ($m = n$ = approximately 8), and “long” ($m = n$ = approximately 18) co-oligomers. For the hexyl series, co-oligomers with varied lipid to guanidinium ratios were also synthesized to determine how this ratio influences siRNA complexation and delivery. Benzyl alcohol, which serves as a useful handle for characterization purposes, was initially chosen as the initiator. As a further demonstration of the rapid tuning and versatility of this strategy, later iterations incorporated the more complex poly(ethylene glycol) methyl ether (PEG), to explore its effect on complexation and biodistribution. Block co-oligomers were synthesized by mixing the requisite initiator first with each one of the hydrophobic monomers **2–5** in the presence of both the thiourea catalyst (TU) and catalytic base (1,8-diazabicycloundec-7-ene, DBU). When the first lipid block had formed, guanidine-protected monomer **6** was added, leading to incorporation of the second protected-guanidine block. Alternatively, statistical co-oligomers were synthesized by using a mixture of the hydrophobic monomer and protected-guanidine monomer at the start of the reaction. Deprotection of the guanidine functionality with trifluoroacetic acid (TFA) in either the block or statistical co-oligomers yielded the desired amphipathic carbonate co-oligomers (Table 1). The synthesized amphipathic carbonate co-oligomers represent variations in overall length (DP = degree of polymerization), initiator moiety, hydrophobic substructure, and hydrophobic to guanidinium ratio, collectively allowing for systematic examination of their ability to noncovalently complex, deliver, and release siRNA.

Characterization of siRNA:Co-oligomer Complexes. As these guanidium-rich amphipathic carbonate co-oligomers represent a new class of delivery vehicles, our initial focus was to determine whether they would spontaneously form complexes with siRNA. A gel shift assay was used for this purpose. To form the siRNA:co-oligomer complexes, a solution of siRNA in phosphate buffered saline (PBS) was added to a solution of co-oligomer in PBS to obtain siRNA:co-oligomer molar ratios of 1:1, 1:5, 1:10, and 1:25. The resulting solutions were incubated at room temperature to allow time for complex formation (30 min). The complexes were then loaded onto an agarose gel, fractionated, and subsequently stained with ethidium bromide. The

Table 1. Representative synthesized guanidinium-rich amphipathic carbonate co-oligomers. Co-oligomer descriptors refer to the hydrophobic side chain (E = ethyl, H = hexyl, D = dodecyl, Chol = cholesterol) and to the guanidinium group (G) used, followed by numbers that reflect the average number of the respective monomers in the oligomer. The n and m refer to Fig. 1C

Co-oligomer number	Co-oligomer name	n	m	R	Initiator	Type	M_w/M_n^*
7a	E:G 5:5	5	5	ethyl	benzyl	block	1.46
7b	E:G 8:9	8	9	ethyl	benzyl	block	1.30
7c	E:G 19:19	19	19	ethyl	benzyl	block	1.42
8a	H:G 4:4	4	4	hexyl	benzyl	block	1.46
8b	H:G 8:9	8	9	hexyl	benzyl	block	1.33
8c	sH:G 9:9	9	9	hexyl	benzyl	statistical	1.26
8d	H:G 10:4	10	4	hexyl	benzyl	block	1.20
8e	H:G 4:10	4	10	hexyl	benzyl	block	1.15
8f	H:G 17:16	17	16	hexyl	benzyl	block	1.35
9a	D:G 4:4	4	4	dodecyl	benzyl	block	1.61
9b	D:G 4.5:5	4.5	5	dodecyl	benzyl	block	1.33
9c	sD:G 4:4	4	4	dodecyl	benzyl	statistical	1.37
9d	D:G 7:7	7	7	dodecyl	benzyl	block	1.47
9e	D:G 18:17	18	17	dodecyl	benzyl	block	1.47
10	PEG-D:G 2.5:3	2.5	3	dodecyl	PEG-2000	block	1.31
11a	Chol:G 3:7	3	7	cholesterol	benzyl	block	1.39
11b	Chol:G 7:7	7	7	cholesterol	benzyl	block	1.37
12	G8	—	8	—	benzyl	block	1.18

* M_w/M_n was determined on the Boc-protected co-oligomer.

ability of each co-oligomer to noncovalently complex with siRNA at a given molar ratio was assessed by the degree to which the migration of the siRNA toward the positive electrode was inhibited (*SI Appendix, Fig. S1*). The highly hydrophobic co-oligomers **8d** and cholesterol-containing **11b** were insoluble in PBS and, therefore, were ineffective in complexing siRNA under the conditions tested. All other co-oligomers with an approximate lipid:guanidinium ratio of 1 or less formed complexes with siRNA.

Gel electrophoresis was also used to assess the hydrolytic stability of the siRNA:co-oligomer complexes. As had been shown previously for the guanidinium-only carbonate oligomers, the carbonate backbone is shelf stable as a solid but, as desired for cargo release after cell entry, it hydrolyses with a half-life of about 8 h in Hepes-buffered saline (pH 7.4, 37 °C) (26). We anticipated that the siRNA:co-oligomer complexes would be similarly stable during cellular entry but subsequently degrade with the release of free siRNA. In a hydrolytic stability assay in the absence of cells, the release of free siRNA could be detected by ethidium bromide staining in a gel shift assay. The siRNA:co-oligomer complexes were incubated for various amounts of time in PBS (pH = 7.4, 37 °C) and then loaded onto a gel and fractionated (*SI Appendix, Fig. S2*). As expected, subsequent staining revealed a stronger uncomplexed siRNA band as the incubation time in PBS increased (8–24 h). While not quantitative, this assay demonstrates that the siRNA:co-oligomer complexes examined in this study generally stay intact during incubation (for at least 4 h), with differences in hydrolytic stability depending on the tunable composition of the co-oligomer. This period is attractive for cell-uptake studies as the complexes are internalized within minutes and before substantial degradation occurs. By 24 h, the complex, regardless of co-oligomer identity, is almost fully degraded. The timing of the siRNA:co-oligomer complex degradation allowed for rapid evaluation of uptake and release in this inaugural study. More generally, this tunable property could also be used to minimize toxicity (see below) or to facilitate local delivery and release, thereby avoiding off-target effects from unintentional systemic exposure.

Dynamic light scattering (DLS) was used to analyze the average diameter of the siRNA:co-oligomer complexes. For these experiments, complexes were formed at the same charge ratio as used in the in vitro siRNA delivery experiments (charge ratio of 4.8/1 +/-) (see below). While all co-oligomers examined by DLS formed complexes whose sizes could be measured immediately upon mixing with siRNA, some of the co-oligomers, includ-

ing a guanidinium-only oligomer, formed aggregates that could not be accurately sized and generally were found to be either less effective or ineffective in delivering siRNA (see below). The size of the siRNA:co-oligomer complexes, which is a tunable function of co-oligomer type and siRNA:co-oligomer ratio, ranged in this study from approximately 200 nm in diameter to approximately 1.5 μ m (*SI Appendix, Table S1*). These are average sizes and by filtration one can obtain smaller (<200 nm) or larger particle sizes. Over the course of the hour measurement period, some of the co-oligomer complexes increased in size, a phenomenon that has been observed previously in the complexation of polynucleotides with oligoguanidiniums (31). Solutions of only the co-oligomer without siRNA did not form measurable particles at the concentrations used for siRNA:co-oligomer complex formation. Size measurements were not optimized for this study, though they can be further modified by varying the ratio of co-oligomer to siRNA by mixing two distinct co-oligomers with one another before mixing with siRNA, or by utilizing PEG initiated co-oligomers (see below). These results demonstrate that the siRNA:co-oligomer complex size can be tuned by modifying the identity of the amphipathic co-oligomer.

Guanidinium-Rich Amphipathic Carbonate Co-oligomer-Mediated siRNA Delivery In Vitro.

The amphipathic carbonate co-oligomers were then screened for their ability to deliver and release siRNA intracellularly. To examine the siRNA:co-oligomer complexes, a dual fluorescent protein reporter assay was used, allowing one protein to be selectively suppressed by its siRNA while the other serves as an internal control. Immortalized human keratinocytes (HaCaTs) were transduced with two distinct lentiviral reporter constructs; one expresses enhanced green fluorescent protein (EGFP), and the other consists of tandem luciferase (Luc2) and tomato fluorescent protein (tdTOM) (Luc2/tdTOM) (32). The siRNA-entitled CBL3 (33) targets the Luc2/tdTOM fusion construct and would therefore reduce, if successfully delivered, the red fluorescent protein expression without affecting the green fluorescent protein expression. The EGFP reporter is valuable as a control for potential nonspecific effects of siRNA administration.

In these experiments, siRNA:co-oligomer complexes were formed by mixing together siRNA and co-oligomer solutions at a positive to negative (+/-) charge ratio of 4.8:1, which was held constant to allow for comparisons of different co-oligomers. The resultant complex was incubated at room temperature for 30 min

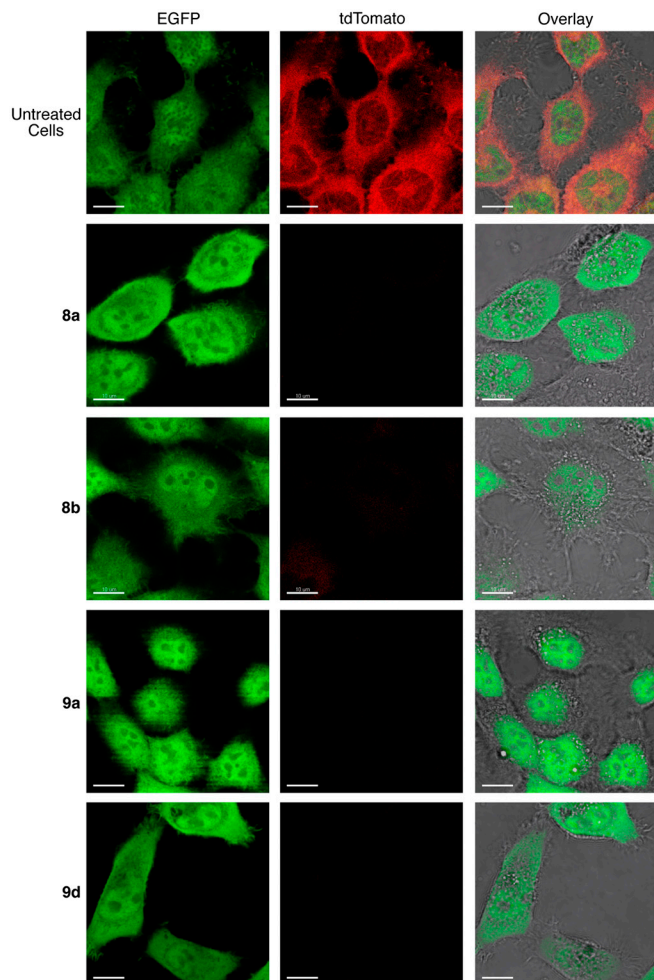


Fig. 3. Fluorescence microscopy of tdTomato (tdTOM) fluorescence and EGFP fluorescence in dual fluorescence reporter HaCaT cells treated with siRNA:co-oligomer complexes. Prior to these experiments, cells had been sorted and >90% of the cells express both tdTOM and EGFP. Treatments were 100 nM with respect to siRNA. The left panel is the green channel only, the middle panel is the red channel only, and the right panel is an overlay of the brightfield image, red channel, and green channel. The scale bar is 10 μ m.

Mixtures of Amphipathic Carbonate Co-oligomers for the Delivery of siRNA into HaCaT Cells and Primary Keratinocytes. In addition to the diversity and, thus, tunability that can be rapidly achieved with this step-economical co-oligomer synthesis strategy, including incorporation of various hydrophobic side chains and initiator

moieties, two or more distinct co-oligomers can also be mixed with one another to obtain even greater diversity in siRNA complexation systems and thus in siRNA delivery (or other properties). To demonstrate this option, PEG-initiated co-oligomer **10** was mixed at various molar ratios with **9a**, and then this mixture was combined with siRNA and the resulting complex applied to the dual fluorescence reporter HaCaT cells (Fig. 4A). PEG-initiated co-oligomer **10** exhibits no siRNA delivery ability; however, as the percentage of PEG-initiated **10** relative to **9a** decreases, knockdown increases toward the values obtained for **9a** alone. Significantly, this strategy allows one to tune the size and stability of the complexes as the DLS of the **10** and **9a** complexes with siRNA show a striking decrease in average size and increase in stability by the addition of the PEG-initiated co-oligomer **10**, with all mixtures of **10** and **9a** exhibiting average sizes below 300 nm (Fig. 4B). The incorporation of PEG into these siRNA:co-oligomer complexes, and the resultant control of size and stability are desirable for transitioning to in vivo experiments and other applications (36).

These smaller and more stable siRNA:co-oligomer complexes were then examined for their ability to deliver siRNA and induce knockdown in primary keratinocytes. Primary keratinocytes, which had been transduced with the EGFP and Luc2/tdTOM plasmids, were treated with a siRNA:co-oligomer complex composed of 95:5 **9a**:**10** in a dose-dependent manner, resulting in up to 79% knockdown, with significant knockdown (62%) at only 10 nM (Fig. 4C).

Cytotoxicity of the siRNA:co-oligomer Complexes. To determine the cytotoxicity of the siRNA:co-oligomer complexes, HaCaT cells were treated with siRNA:co-oligomer complexes and then analyzed colorimetrically for mitochondrial reduction of methylthiazolyl-diphenyltetrazolium bromide (MTT) relative to untreated cells (SI Appendix, Fig. S6). While this study was designed to explore complexation, uptake, and release and not to minimize toxicity, most of the amphipathic co-oligomer complexes were relatively nontoxic to HaCaT cells at the tested concentrations. Some co-oligomer complexes did display some cytotoxicity but only at the highest concentrations tested (100 nM with respect to siRNA). Importantly, lowering the dose from 100 nM to 50 nM substantially reduces the cytotoxicity of even the most active co-oligomer complexes, conditions which only slightly attenuate the knockdown efficacy. In addition, exchange of the counterion from TFA to chloride reduces cytotoxicity without affecting knockdown efficiency.

Further underscoring the value of biodegradable release, when the siRNA:co-oligomer complexes were pre-incubated in PBS (pH = 7.4, 37°C) for 24 h (conditions which were shown by gel electrophoresis to result in degradation of the complex), no

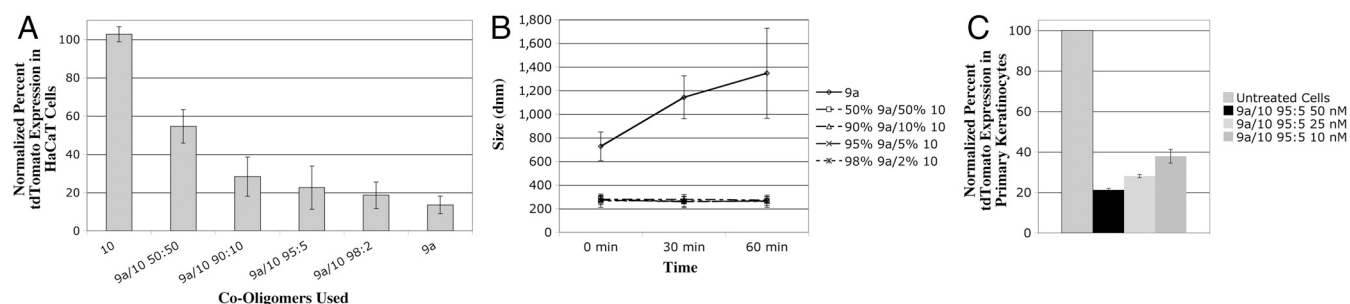


Fig. 4. tdTOM reduction in HaCaT cells and primary keratinocytes by and DLS of siRNA:co-oligomer complexes made from various molar percentage mixtures of **9a** and **10**. The percentages are molar percentages out of a total molar ratio of 1:52.5 siRNA:co-oligomer. (A) Co-oligomers **9a** and **10** were mixed at various molar ratios and then mixed with siRNA. The resulting complexes were applied to cells and analyzed by flow cytometry. Results are of at least three separate experiments; each condition in triplicate. Error bars indicate SD. (B) DLS data on the siRNA:co-oligomer complexes made from mixtures of **9a** and **10** with siRNA. The value shown is the average of three separate trials; error is SD. (C) Primary keratinocytes were treated with siRNA:co-oligomer complexes composed of 95:5 **9a**:**10** in a dose-dependent manner. The value shown is the average of three separate trials. Error is SD.

